







## Respiratory Protective Equipment: Respiratory Protection Program

- Administration
- Knowledge of Respiratory Hazards
- Assessment of Respiratory Hazards
- Control of Respiratory Hazards
- Selection of Proper Respiratory PPE
- Training
- Inspection, Maintenance, Repair of Equipment
- Medical Surveillance

## Respiratory Protective Equipment: Types of Hazards

- **Gas & Vapor Contaminants**
- Inert non-reactive (helium, neon, argon)
- Acidic pH < 7 (HCl, SO<sub>2</sub>) *sour*
- Alkaline pH > 7 (NH<sub>3</sub>) *bitter*
- Organic compounds of Carbon (CH<sub>4</sub>)
- Organometallic metals + organics; Pb(C<sub>2</sub>H<sub>5</sub>)<sub>4</sub> (tetraethyl lead)
- Hydrides hydrogen + metals;  $B_2H_6$  (diborane)

# Respiratory Protective Equipment: Types of Hazards

### **Particulate Contaminants**

- Dust solid, mechanically produced particle
- Spray liquid, mechanically produced particle
- Fume solid condensation particle
- Mist fine liquid condensation particle
- Vapors very fine particle from volatilized liquid
- Fog mist of sufficient [c] to obscure vision
- Smoke solid/liquid gas combo that obscures vision

### Respiratory Protective Equipment: Selection Criteria

- Nature of hazardous operation/process
- Type of respiratory hazard
- Proximity of hazardous area to non-hazardous area
- Period of time that respiratory protection necessary
- Activities of worker in hazardous area
- Functional and physical characteristics of various types of respirators
- Respirator-protection factor and respirator fit

Respiratory Protective Equipment: Classes of Devices

Class 1 – Air-purifying
 Class 2 – Atmosphere or Air-supplying
 Class 3 – Combination Air-purifying and Atmosphere-supplying

Note: All above-mentioned respirators must be approved by NIOSH-MSHA

Mechanical-filter
Chemical-cartridge
Combination mechanical/chemical
Gas masks
Powered air-purifying

### **Mechanical-filter**

- Protects against airborne particulate matter only
- Quarter, half, and full face models available
- Has fibrous material that traps particles
- Specialization emphasized depending on use

### **Chemical-cartridge**

- Protects against certain airborne gases and vapors only
- Has chemical cartridge to remove hazards
- Non-emergency protective devices; never use in IDLH atmospheres

# Respiratory Protective Equipment: Class 1 – Air-purifying respirators Chemical-cartridge Do NOT use when:

- Gaseous material is extremely toxic in very small concentrations
- Harmful gaseous material cannot clearly be detected by odor

Gaseous material is present in [c] highly irritating to the eyes w/o proper eye protection

 Gaseous material is not effectively stopped by chemical fills used, regardless of concentrations

Combination mechanical/chemical
Utilizes both types of filters for atmospheres with both particulates, gases, and vapors
Filters can be combination all-in-one or independently replaceable

### Gas masks

- Designed solely to remove specific contaminants from air using a canister
- May be used for *escape-only* from IDLH atmospheres only, not entrance
- Suitable for ventilated areas that are *not* subject to rapid changes in air-contaminant levels
- Canisters come in three sizes: supersize, industrial size, chin-style

### Gas masks

- Replace canister when:
  - Canister with window indicator shows specified color change
  - Leakage detected by smell, taste, or eye, nose, or throat irritation
  - High resistance to breathing develops
  - Canister shelf life exceeded
  - Uncomfortable heat in inhaled air experienced
  - Wearer has feeling of nausea, dizziness, or ill-being

### **Powered air-purifying**

- Utilizes any one or all types of filters, cartridges, combos or canister for atmospheres with particulates, gases, and vapors present individually or as a group
- Uses a power source (battery pack) to operate a blower that passes air across a filter
- Supplies air at positive pressure so any leakage is outward from facepiece

Supplied-air (SAR)
 Self-contained Breathing Apparatus (SCBA)
 Combination-SCBA and Supplied-air

Respiratory Protective Equipment: Class 3 – Combination Airpurifying and Air-supplying

**Combination-SCBA and Supplied-air** 

- Combination air-line respirator w/ auxiliary airpurifying attachment
- Air-line respirator must have respirable air
- Limitations:
  - ♦ Use in filtering mode for escape only
  - Not for use in IDLH atmospheres

Not for oxygen-deficient atmospheres (<19.5%)</p>

Use only approved hose lengths and pressure ranges

♦ When airflow is cut, switch to filter and exit to clean air

## Respiratory Protective Equipment: Decision Logic

- Skin Absorption (SCBA if injury or death possible)
- Warning Properties (odor, taste, irritation < PEL)
- Sorbent Efficiency (>3 minute breakthrough)
- **Eye Irritation** (Use full face respirator if issue)
- IDLH (escape possible <30 minutes without death/permanent health damage)</p>
- Lower Flammable Limit & Firefighting (maximum protection pressure-demand SCBAs needed because of IDLH environment)

## Respiratory Protective Equipment: Assigned Protection Factors

APF 5 **APF 10 APF 25 APF 50 APF 100** APF 1,000 APF 2,000 APF 10,000

- quarter-face respirator (QF)
- half-face respirator (HF)
- ~ powered, air purifying (PAP)
- full-face respirator (FF)
- ~ PAP/HEPA or SA/constant
- supplied air/pressure +/-
- supplied air/FF/pressure +/-
- SCBA/FF/pressure +/-

## **Respiratory Protective Equipment: Assigned Protection Factors**

### Assigned Protection Factor Reference Guide\*

Type of Respirator	ANSI	OSHA APFs N														NIOSH		
	Z88.2-1992	As	Pb	Pb-C	Cd	Bz	Fm	Ar	COV	CD	DBCP	AN	EtO	VC	MDA	BD	MC	RDL
<u>Air Purifying</u> Single Use, Quarter Mask Half Mask, incl. Disposables Full Facepiece	10 10 100	10 50	10 50	10 50	10 50	- 10 50	10	10 50	10 10	5 10 100	-	10 50	50	10 25	10 50	10 50	-	5 10 50
<u>Powered Air Purifving</u> Half Mask Full Facepiece Helmet/Hood Loose-Fitting Facepiece (ANSI)	50 1,000 1,000 25	100 100 100 100	1,000 1,000 1,000 1,000	50 50 25 25	50 250 25 25	100 -	-	1,000 1,000 1,000 1,000	>10 >10 >10 >10 >10	>100 >100 >100 >100 >100	-	-	-	25 25 25 25	1,000 - -	50 50 25 25	-	50 50 25 25
<u>Continuous Flow</u> Half Mask Full Facepiece Hood/Helmet Loose-Fitting Facepiece (ANSI)	50 1,000 1,000 25	100 100 100 100	1,000 2,000 2,000 2,000	50 50 25(3) 25	50 250 25 25	1,000	- 100 100 100	10 50 2,000 2,000	>10 >10 >10 >10 >10	-	1,000 2,000 2,000 2,000	10 50 2,000 2,000	2,000 2,000 2,000 2,000	1,000 1,000 1,000 1,000	-	25 25	25 200(2) 25 25	50 50 25 25
<u>Pressure-Demand</u> Half Mask Full Facepiece	50 1,000	1,000	1,000 2,000	1,000 2,000	1,000 1,000	1,000	-	1,000 2,000	>10 >10	-	1,000 2,000	2,000	2,000	-	-	1,000 1,000	200(2)	1,000 2,000
Self-Contained Breathing App. Pressure Demand Open/Closed Circuit Full Facepiece	(1)	>1,000	>2,000	>2,000	>1,000	>1,000	>100	>2,000	>10	-	> 2,000	>2,000	>2,000	>3,600	>1,000	>1,000	>200(2)	10,000

(1) Normally 10,000, but recent studies have indicated that all users may not achieve protection factors of 10,000. For emergency planning purposes where hazarchous concentrations can be estimated, an APF of no higher than 10,000 should be used. (2) For MC, 200xPEL=IDLH Concentration (5,000 ppm).

(3) OSHA has granted an APF of 1,000 to the Bullard models 77 and 88 abrasive blasting respirators for use under the Lead in Construction Standard.

### OSHA APEs:

- GI: General industry, i.e. substances for which there is no substance-specific standard covering respirator selection. Ashestos 1910.1001 As:
- Lead in General Industry 1910.1025 Pbc

- Pb-C: Lead in Construction 1926.62 Cd: Cadmium 1910.1027
- Bz Benzene 1910.1028
- Fmc Formaldahyda 1910.1048
- Ar: Arsanic 1910.1018
- COV: Coke Dvan Emissions 1910.1029 CD: Cotton Dust 1910.1043 DBCP: 1,2-Dibromo-3-Chloropropane 1910.1044 AN: Acrylonitrile 1910.1045 EtD: Ethylana Oxida 1910.1047
- VC: Vinyl Chlorida 1910.1017 MDA: Methylanedianline 1910.1050 1.3-Buradiane 1910.1051 BD: MC: Mathylane Chloride 1910.1052



877-BULLARD (285-5273) www.bullard.com

\* This guide is a compliation of Assigned Protection Factors published by various governmental agencies and the American National Standards Institute.

\*\* General Industry numbers are not currently published. The numbers listed on this chart are the numbers that OSHA enforces to the best of our knowledge. For this reason, be sure to consult 29 CFR 1910.134, other established regulatory standards and/or your local OSHA representative. The Assigned Protection Factors herein do not constitute a recommendation by Bullard for the use of any respirator herein for any hazard.

### Respiratory Protective Equipment: Filter Selection

- Selection of N-, R-, and P-series filters depends on the presence or absence of oil particles, as follows:
- If no oil particles are present, use any series (N, R, or P)
- If oil particles are present, use only R or P series
- If oil particles are present and the filter is to be used for more than one work shift, use only P series

### Respiratory Protective Equipment: Filter Selection

N for Not resistant to oil
R for Resistant to oil
P for oil Proof

Selection of filter efficiency (i.e., 95%, 99%, or 99.97%) depends on how much filter leakage can be accepted.

Examples: N-95 for TB, P-100 for ER, etc.

### **SCBA** Respirator





### CBRNAgent Approved

See Instructions for Required Component Part Numbers, Accessories, and Additional Cautions and Limitations of Use

•Look to see if the CBRN Agent Approval label shown below is on the respirator. If an SCBA is CBRN-approved by NIOSH, it will always carry this label. If this CBRN Agent Approval label is not on the SCBA, the device is not approved by NIOSH for use by emergency responders in CBRN environments. **Check the Label!** 

http://www.cdc.gov/niosh/npptl/cbrncheck.html

### **SCBA Respirator**





### CBRNAgent Approved

See Instructions for Required Component Part Numbers, Accessories, and Additional Cautions and Limitations of Use •The approval number for an SCBA approved for CBRN environments always includes a **CBRN** suffix (TC-13F-XXXCBRN). If the approval number does **not** include a CBRN suffix, it is **not** certified by NIOSH for use by emergency responders in CBRN environments.

•The complete CBRN SCBA assembly must be composed of **only** those component parts listed in the row with the CBRN approval number. Part numbers that are found **only** in the rows of the **non-CBRN** approvals **must not** be used as part of a CBRN SCBA assembly.

### **APR** Canister

Sample abbreviated label below shows the structure of a CBRN canister label.

### Color of the label must be olive.

This label must be attached to the canister. If the label does not fit on the canister, the portion of the label below the dashed line may be provided in a form that is removable, as long as the Cautions and Limitations in the boxed area are added. Dashed lines are added here for illustration only and should not appear on production labels.



•Look to see if the full canister label is like the one shown in Figure 1. Canister labels will specify the CBRN protection level (CBRN Cap 1, CBRN Cap 2 or CBRN Cap 3).

•If an APR is CBRN-approved by NIOSH, the full canister label will identify the CBRN protection level, part number and if the canister has been approved for use for CBRN protection with an approved CBRN APR facepiece. If the full canister label does **not** identify the above information, the device is **not** approved by NIOSH for use by emergency first responders in CBRN environments.

### http://www.cdc.gov/niosh/npptl/cbrnaprcheck.html

### **APR** Canister

G		XYZ Safety Company ANYWHERE USA 1-800-XXX-XXXX XYZ AIR-PURIPYING GAS MASK CANISTER THIS CANISTER IS APPROVED BY YITH HERCLOWING COMPREMENTION 2													
TC-	PROTECTION <sup>1</sup>	CAMISTER	ALTERNATE FACEPIECE			Connection & Gaskiet	SHROUD (CBRN)	NDECUP	NDEECLP	ACCESSORIES		RIES	CAUTIONS AND LIMITATIONS 2/3		
		CBRNCap 1	Future, small	Future, medum	F uture, terge	40 MM Connector	Bulyl Costed Nyon Broudthood	Nosecup grieß	Nosecup med / large	Spectacle Kit	Eye Lens Outset	Spectacle Kit			
		12946002	7/0/25	7,0724	7,0728	17880t/Z	17680trZ	71663672	71560072	2,408084	777542	2,001009			
14G-30000	CBRN Cap 1	X	х	х	х	х	х	х	х	х	х	х	AJUNOSR TV/WYZ, NH, QQ, UU		

1. PROTECTION

CERN Cap 1 - Capadity meets minimum 15 minutes test time

CBRN - Cherrical. Bolonical. Radiological and Nuclear

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### 1. CBRN CAUTIONS AND LIMITATIONS

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104When used all defined occupational exposure limits, the mixed service time current be acceeded. Follow exhibitived currieter change out schedule or observe limit of Service Life indicators to ensure that currieters are replaced before breakthrough occurs.

QQ-Use in conjunction with personal protective ensembles that provide appropriate levels of protection against dermal hazard Palare to do so may result in personal injury even when the respirator is properly titled, used, and maintained.

U.J-The resolution the used beyond eight (0) hours after trillal econate to chemical warfare acents to world oceability of acen permeation. If light exposure is encountered, the respirator should not be used for more than two (2) hours. •Look to see if the full canister label is like the one shown in Figure 1. Canister labels will specify the CBRN protection level (CBRN Cap 1, CBRN Cap 2 or CBRN Cap 3).

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### **APR** Respirator

4	XYZ Salety Company ANYWHERE USA 1-800-XXX-XXXX												NIOSH	
			s RD	x	YZ A	in-P	URIP PROV		GA	S MA	SK I	CWN	IRATOR	
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10-1333	CERN Cap 1	X	х	х	x	x	1	1	X	х	х	х	A A MORRTYWEYZ, HH, GD, UU	1

CERN Cap 1 - Capacity men nationum 15 minutes test tim

CERN - Chemical Biological Radiological and Nuclear

### THE AND UNITED IN

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http://www.cdc.gov/niosh/npptl/cbrnaprcheck.html

### Weapons of Mass Destruction (WMD)







### Potential Types of WMD

Biological – Anthrax, Smallpox, Botulism, Plague, Tularemia, Viral Hemorrhagic fevers, Brucellosis, Ricin, etc.

Chemical – Nerve Gases (VX), Blister/Vesicants (Mustard Gas), Pulmonary Damaging (Phosgene, Chlorine), etc.

Radiological – "Dirty" Bombs

### Biological Types of WMD



**Biological Types of WMD** Category A  $\diamond$  Anthrax ♦ Smallpox ♦ Botulism ♦ Plague ♦ Tularemia ◆ Viral Hemorrhagic fevers (e.g., Ebola) ■ Category B ♦ Brucellosis ■ Other ♦ Ricin

## Biological WMD – CDC Category A

- Easily disseminated or transmitted from person to person
- Result in high mortality rates and have potential for major public health impact
- Might cause public panic and social disruption
- Require special action for public health preparedness

**Biological WMD - Anthrax** Early Symptoms: Cutaneous (skin) - boil-like lesion that becomes ulcer with a black center ◆ Inhalation - flu-like symptoms Ingestion - acute inflammation of intestinal tract Cause: bacterium Bacillus anthracis ■ Not contagious person to person (casual) ■ Treatment: ♦ Ciprofloxacin ♦ Doxycycline



**Biological WMD - Smallpox** Early Symptoms ◆ high fever, fatigue, head and back aches occur initially Prominent Symptoms: Characteristic rash, most prominent on face, arms, and legs follows in 2-3 days. Rash starts with flat lesions that evolve at same rate Cause: variola virus *Contagious person to person (respirated saliva)* Treatment: ♦ supportive care Cidofovir (in vitro)

**Biological WMD - Botulism** Early Symptoms (foodborne) nausea, dry mouth, vomiting, abdominal cramps, and diarrhea Prominent Symptoms: Blurred or double vision, difficulty swallowing and speaking, weakness and/or paralysis first in the arms and then the legs can follow Cause: Clostridium botulinum bacterium toxin Not contagious person to person **Treatment:** ♦ supportive care ♦ Trivalent equine antitoxin

### Biological WMD – Plague

### Early Symptoms:

- (bubonic) fever, chills, muscle aches, headache, nausea, vomiting, diarrhea, abdominal pain, and extreme exhaustion; swollen and tender near infected flea bite
- (pneumonic) pneumonia-like symptoms
- Cause: bacterium Yersinia pestis
- Bubonic Not contagious person to person
- Pneumonic contagious person to person (sneeze/cough)
- Treatment:

bubonic – widely available antibiotics
 pneumonic – Streptomycin, Gentamicin



# Biological WMD – Tularemia

### Early Symptoms:

- Skin- Presence of skin ulcer at location where bacteria entered skin and swollen glands
- Ingestion throat infection, abdominal pain, diarrhea, and vomiting
- Inhalation pneumonia-like illness
- Eyes painful swelling of lids, red eyes, visual pain
- Cause: bacterium *Francisella tularensis*
- Not contagious person to person
- Treatment:
  - ♦ Streptomycin
    ♦ Gentamicin


# Biological WMD – Viral Hemorrhagic Fevers

#### Early Symptoms:

 Overall vascular system is damaged, body's ability to regulate itself is impaired, often accompanied by hemorrhage (bleeding)

■ Causes:

 Arenaviruses (Lassa, Machupo); Filoviruses (Ebola, Marburg); Bunyaviruses; Flaviviruses

Can be contagious, depending on type

Treatment:

Supportive care
 Lassa fever - Ribavirin



# Biological WMD – CDC Category B

- Moderately easily to disseminate or transmit from person to person
- Result in moderate morbidity rates and low mortality rates
- Require specific enhancements of CDC'c diagnostic capacity and enhanced disease surveillance

# Biological WMD – Brucellosis (Category B)

Early Symptoms: similar to flu; may include fever, sweats, headaches, back pains, and physical weakness ■ Cause: bacteria of genus *Brucella* Not contagious person to person ■ Treatment: ♦ Doxycycline Streptomycin

♦ Rifampin



## Biological WMD – Ricin

#### Early Symptoms:

- Inhalation: coughing, tightness in the chest, difficulty breathing, nausea, and aching muscles.
- Ingestion: internal bleeding of the stomach and intestines leading to vomiting and bloody diarrhea.
- Injection: muscles and lymph nodes near the injection site would die.
- Cause: poison from processing castor beans
- Not contagious person to person
- **Treatment:** 
  - Supportive care



## Chemical Types of WMD



## Brief History of Chemical Warfare

- April 1915 World War I (Ypres, Belgium) Over 5,000 allied troops died in first chlorine gas attack; 15,000 wounded
- World War I British and German forces used chorine gas, mustard gas (July 1917), and phosgene (late 1915) heavily before the war ended.
- World War I Overall about 113,000 tons of chemical weapons were used; killing around 92,000; total of 1.3 million casualties.
- 1925 Geneva protocol ("Protocol for the prohibition of the use in war of asphyxiating, poisonous or other gases, and of bacteriological methods of warfare") was signed by the league of nations (initial signing 38 nations; now over 130 nations (manufacture and threat of use of chemical weapons not prohibited; protocol is vague on "other gas;" and punishment for use

http://www.geocities.com/CapeCanaveral/Lab/4239/chemweapons/history.html

# Brief History of Chemical Warfare

- World War II No chemical weapons used in battle; large amounts of new chemical weapons (nerve agents) developed in 1930's & 1940'S by the Germans were discovered
- 1952 United Kingdom discovers VX based on insecticide research.
- **1961** Manufacturing of VX begins in US (Ends in 1968)
- Vietnam War Chemical Agents ("Agent Orange", "Agent Purple", "Agent Blue", and "Agent White", were used to defoliate vegetation surrounding the enemy.
- 1980's Chemical warfare agents use reported in Laos, Cambodia, Afghanistan, Iran, and Iraq ("Gulf war Syndrome")
- March 1995 Tokyo, Japan Religious cult releases form of sarin gas in Tokyo's subway system during morning rush hour; attack killed 11 people and injured over 5,500 people (trample).

http://www.geocities.com/CapeCanaveral/Lab/4239/chemweapons/history.html

## **Clues of Chemical Terrorism**

- Unusual increase in number of people seeking care (respiratory, neurological, gastrointestinal symptoms)
- Symptom clustering, unusual age distribution
- Location of release not consistent with chemical's use
- Simultaneous impact to human, animal, and plant populations
- Unusual clustering of patients in time or location

## Chemical Types of WMD

Blister/Vesicants (Lewisite, Mustard Gas) Blood (Arsine, Hydrogen Cyanide) Pulmonary Damaging (Chlorine, Phosgene) ■ Incapacitating (BZ, LSD) ■ Nerve (Sarin, VX) Riot Control/Tear (Bromobenzylcyanide) ■ Vomiting (Adamsite) Other Industrial Chemicals

## Chemical – Blistering/Vesicant Names: ◆ Lewisite (L), ◆ Mustard gas (H),



- ◆ Nitrogen mustard (HN-1, HN-2, HN-3),
- Mustard/Sulfur mustard (HD, H),
- Phosgene oxime (CX)
- **Effects:** 
  - ◆ Severe irritation immediately after exposure
  - Causes skin blisters and redness
  - Causes tearing, conjunctivitis, corneal damage
  - Mild respiratory distress to marked airway damage
  - ♦ May cause death

Chemical – Blistering/Vesicant Unique Characteristics: Mustard – burning garlic/horseradish odor Lewisite – penetrating geranium odor Phosgene oxime – pepperish/pungent odor Decontamination: Immediately remove clothing ◆ Gently wash skin with soap and water  $\diamond$  Do not abrade skin For eyes, flush with plenty of water or normal saline

#### Chemical – Blistering/Vesicant First Aid: Immediately decontaminate skin ◆ Flush eyes for 10-15 minutes with water/saline ◆ If breathing difficulty, give oxygen ♦ Supportive care Other Considerations: Mustard has asymptomatic latent period No antidote or treatment for mustard

♦ Lewisite – immediate burning pain, blisters later

Antidote: injection of British Anti-Lewisite (BAL)

Phosgene oxime – immediate pain

Possible pulmonary edema

Chemical – Asphyxiant/Blood ■ Names: ♦ Arsine Cyanogen chloride Hydrogen cyanide ■ Effects: Confusion ♦ Nausea ◆ Gasping for air, but more abrupt onset Seizures prior to death

Chemical – Asphyxiant/Blood Unique Characteristics: Possible cherry red skin Possible cyanosis (bluish skin) Possible frostbite (liquid arsine, cyanogen chloride) ◆ *Arsine* - has garlic-like or fishy smell at high concentrations **Decontamination:** ◆ Immediately remove clothing if no frostbite ♦ If frostbite, wash with warm water to release clothing Gently wash skin with soap and water Do not abrade skin ♦ For eyes, flush with plenty of water or normal saline

## Chemical – Asphyxiant/Blood

First Aid:

Rapid treatment with oxygen

For cyanide

- Antidote: Sodium nitrite, then sodium thiosulfate

Other Considerations:

 Arsine and cyanogen chloride may cause delayed pulmonary edema (fluid buildup)

Chemical – Choking/Pulmonary ■ Names: ♦ Chlorine Hydrogen chloride Nitrogen oxides ♦ Phosgene ■ Effects: Eye and skin irritation ♦ Airway irritation  $\diamond Dyspnea$  (labored breathing), cough ♦ Sore throat Chest tightness

#### Chemical – Choking/Pulmonary

Unique Characteristics:

- Chlorine greenish-yellow gas; pungent odor
- Phosgene newly-mown hay or grass odor
- Possible frostbite (liquid phosgene)
- Decontamination:
  - Immediately remove clothing if no frostbite
  - ◆ If frostbite, wash with warm water to release clothing
  - ♦ Gently wash skin with soap and water
  - Do not abrade skin
  - For eyes, flush with plenty of water or normal saline

## Chemical – Choking/Pulmonary ■ First Aid: Fresh air, forced rest Semi-upright position ◆ If respiratory distress, oxygen w/wo positive airway pressure may be needed $\diamond$ Other supportive therapy, as needed Other Considerations: May cause delayed pulmonary edema, even following symptom-free period (duration dependent on dose)

#### Chemical – Nerve

Names:
Cyclohexyl sarin (GF)
Sarin (GB)
Soman (GD)
Tabun (GA)
VX

Effects:

Miosis (pinpoint pupils), blurred/dim vision
 Headache

♦ Nausea, vomiting, diarrhea

Copious secretions/sweating

Muscle twitching/fasciculations, seizures

Breathing difficulty

#### Chemical – Nerve

Unique Characteristics: Miosis (pinpoint pupils) Copious secretions/sweating Muscle twitching/fasciculations Tabun - slightly fruity odor; Soman - slight camphor odor Decontamination: Immediately remove clothing ♦ Gently wash skin with soap and water Do not abrade skin ◆ For eyes, flush with plenty of water or normal saline

#### Chemical – Nerve

■ First Aid: Atropine before other measures Pralidoxime (2-PAM) chloride Other Considerations: Onset of symptoms from dermal contact with liquid forms may be delayed Repeated antidote administration may be necessary

Chemical – Incapacitating/Behavior ■ Names: ◆Agent 15/BZ ■ Effects: ◆ Dry mouth and skin ◆ Initial tachycardia (fast heartbeat) Altered consciousness, delusions, belligerence ♦ Hyperthermia ◆ Ataxia (lack of coordination) ♦ Hallucinations  $\diamond$  Mydriasis (dilated pupils)

Chemical – Incapacitating/Behavior Unique Characteristics: Mass drug intoxication, hallucinations Hyperthermia Mydriasis (dilated pupils) Decontamination: ♦ Immediately remove clothing ♦ Gently wash skin with soap and water  $\diamond$  Do not abrade skin

Chemical – Incapacitating/Behavior ■ First Aid: Remove heavy clothing Evaluate mental status ◆ Use restraints as needed Monitor core temperature carefully ◆ Supportive care Other Considerations: Hyperthermia and self-injury are largest risks  $\diamond$  Hard to detect – odorless, non-irritating Possible serious arrhythmias (erratic heartbeat) Antidote - physostigmine

## Chemical WMD – Using Colorimetric Devices





## Colorimetric – Nerve Agents



Smart M-8 Nerve Agent Detector

- Instantly identifies nerve (V, G, H) agents with the latest technology.
- Includes clip and peel and stick back.
- Color indicator on each individual card.

http://www.smart-strip.com/order.htm

## Colorimetric – HAZMAT

	, E	FLUORIDE	NERVE
		100	
		100	
		c	YANIDE
OXIDIZER ARE	ENIC S	HAZO	AT
	11.01.01.004	SMART S	tink-

#### **HazMat Smart-Strip**

- Instantly identifies nerve, cyanide, hydrogen sulfide, arsenic, acids/caustics, fluoride, oxidizers, and chlorine.
- (12 hour operational timeframe)
- 2 year shelf life.

http://www.smart-strip.com/order.htm

## Colorimetric – Civil Defense



Simultaneous Kits for Domestic Preparedness

Draeger has developed an important new detection system that detects up to five different chemical agents simultaneously in just 5 minutes.



http://www.afcintl.com/kits1.htm

## Chemical WMD – Using PIDs





## Using PIDs – Function

- A Photo Ionization Detector (PID) uses an Ultraviolet (UV) light source (Photo= light) to break down chemicals to positive and negative ions (Ionization) that can easily be counted with a Detector.
- Ionization occurs when a molecule absorbs the high energy UV light, which excites the molecule and results in the temporary loss of a negatively charged electron and the formation of positively charged ion.

http://www.afcintl.com/pdf/pid.pdf

## Using PIDs – Function

- The gas becomes electrically charged. In the Detector these charged particles produce a current that is then amplified and displayed on the meter as "ppm" (parts per million) or even in "ppb" (parts per billion).
- The ions quickly recombine after the electrodes in the detector to "reform" their original molecule.
   PIDs are non-destructive; they do not "burn" or permanently alter the sample gas, which allows them to be used for sample gathering.
- PID calibrated with isobutylene (benzene in past)

http://www.afcintl.com/pdf/pid.pdf

## Using PIDs – Function



## Using PIDs - Measurement

- All elements and chemicals can be ionized, but they differ in the amount of energy they require.
- The energy required to displace an electron and "ionize" a compound is called its Ionization Potential (IP), measured in electron volts (eV).
- The light energy emitted by an UV lamp is also measured in eV.

## Using PIDs - Measurement

- If the IP of the sample gas is less than the eV output of the lamp, then the sample gas will be ionized.
- If the "wattage" of a gas or vapor is less than the "wattage" of the PID lamp, then the PID can "see" the gas or vapor.
- If the "wattage" of the gas or vapor is greater than that of the PID lamp the PID cannot "see" the vapor.

### Using PIDs – Measurement

Some Ionization Potentials (IPs) for Common Chemicals



## Using PIDs - Detectables

**Organics:** compounds *containing* Carbon (C) atoms:

- **Aromatics** benzene ring (benzene, toluene, xylene)
- Ketones & Aldehydes C=O (acetone, MEK)
- Amines & Amides nitrogen (diethylamine)
- Chlorinated hydrocarbons TCE, PERC
- Sulfur compounds mercaptans, sulfides
- Unsaturated hydrocarbons butadiene, isobutylene
- Alcohols- isopropanol (IPA), ethanol
- Saturated hydrocarbons butane, octane

**Inorganics:** Compounds *without* Carbon atoms:

 Ammonia, Arsine, Phosphine, Hydrogen Sulfide, Nitric Oxide, Bromine, and Iodine

http://www.afcintl.com/pdf/pid.pdf
### Using PIDs – Non-Detectables

- Radiation
- Air  $(N_2, O_2, CO_2, H_2O)$
- Common Toxics (CO, HCN, SO<sub>2</sub>)
- Natural Gas (Methane, Ethane)
- Acid Gases (HCl, HF, HNO<sub>3</sub>)
- Others- Freons, Ozone (O<sub>3</sub>), Hydrogen peroxide
- Non-volatiles: PCBs, Greases

### Using PIDs – eV Bulb Tips

9.8 & 10.6 eV versus 11.7 eV PID Lamps

- 9.8 and 10.6 are more specific
- 9.8 and 10.6 last a few years
- 9.8 and 10.6 are more sensitive
- 11.7 have a shorter life than 9.8 or 10.6
- 11.7 eV bulbs should only be used when compounds with IPs over 10.6 eV are expected: Examples include methylene chloride, chloroform, and carbon tetrachloride

### Chemical WMD – Monitoring



**Chemical WMD – Monitoring Important Terminology:** CWA – Chemical Warfare Agents **TIC – Toxic Industrial Chemicals TIM – Toxic Industrial Materials** ■ SAW – Surface Acoustical Wave ■ IMS – Ion Mobility Spectroscopy

Wrenn Christopher, <u>Using PIDs in Terrorist Chemical Attacks</u>, Compliance Magazine, February, 2004. Chemical WMD – Monitoring
Important Terminology (military CWA):
SAW – Surface Acoustical Wave – technique in which a high frequency signal, instead of an optical beam, is used to identify substances

IMS – Ion Mobility Spectroscopy technique that measures vibrational frequencies and the bonds between different nuclei, usually to examine ions of relatively low molecular weight and high ion mobility

**Chemical WMD – Monitoring Important Concepts:** "Toxic chemicals in general industry are *more accessible* to terrorists than chemical warfare agents" "The difference between a HAZMAT" incident a terrorist event is intent – accident vs. intentional

Wrenn Christopher, <u>Using PIDs in Terrorist Chemical Attacks</u>, Compliance Magazine, February, 2004.

**Chemical WMD – Monitoring** General Industry Agents ◆ Chlorine – water treatment Phosgene gas – various chemical processes ♦ Parathion – agricultural insecticide ♦ TDI (toluene diisocyanate) – catalyze urethanes Chemical Warfare Agents Chlorine gas Phosgene gas ◆ **Sarin** (Tokyo - 37% sarin, 63% acetonitrile) Lewisite/Mustard gas mixture Wrenn Christopher, Using PIDs in Terrorist Chemical Attacks, Compliance Magazine,

February, 2004.

**Chemical WMD – Monitoring** Low VP Chemicals that Fool IMS algorithms ♦ Brake fluid Diesel additives Paint fumes Glycol ethers/Vinyl esters - cleaning products Wintergreen/spearmint oils – mouthwash/mints PIDs can measure both CWAs and TICs that are outside of CWA SAW/IMS detectors Multiple CWA-specific devices are required to account for cross-sensitivity and reliable determination of CWA

Wrenn Christopher, <u>Using PIDs in Terrorist Chemical Attacks</u>, Compliance Magazine, February, 2004.

### Chemical WMD – Monitoring



### Chemical WMD – Monitoring

Table 1								
Compound	Structure	m.w.	Vapor Press. (ppmv)	Lamp (eV)	CF	8-h TWA (mg/m <sup>3</sup> )	8-h TWA (ppbv)	LC50 (ppmvmin.)
Arsine (SA)	AsH,	78	Gas	10.6	1.9	0.16	0.05	
Cyanogen Chloride	CICN	61.5	1	ND**	ND**	0.6 05	300 Cš	
DMMP	O=P(Me)(OMe),	124		10.6	4.3			
GF	O=PF(Me)(O-Cyclohex)	180		10.6	~3*			
Hydrogen Cyanide	HCN	27	Gas	ND**	ND**	11	10,000	270
Lewisite	CICH=CHAsCI,	207	460	10.6	~1*	0.003	0.35	140
Methyl salicylate	2-(H0)C <sub>6</sub> H <sub>4</sub> C0 <sub>9</sub> Me	152		10.6	0.9			
Mustard (HD)	S(EtCI),	159	95	10.6	0.6	0.003	0.46	
N Mustard (HN-1)	N(Et)(EtCl),	172		10.6	~1*		11	
Phosgene	0=CCI,	99	Gas	11.7	~2*	0.4	100	2
Sarin (GB)	0=PF(Me)(OiPr)	140	3800	10.6	3	0.0001	0.017	-12
Soman (GD)	O=PF(Me)(OCH(Me)(tBu))	182	530	10.6	~3*	0.00003	0.004	9
Tabun (GA)	O=P(CN)(OEt)(NMe <sub>2</sub> )	162	48	10.6	0.8	0.0001	0.015	20
Triethyl phosphate	0=P(OEt)	182		10.6	3.1			
VX	O=P(Me)(OEt)(SetN(iPr)2)	267	0.92	10.6	~0.5*	0.00001	0.00091	2.7

\* Estimated value. \*\* ND = Not Detectable by PID. C = ceiling value.%

#### Low CF = High PID Sensitivity To A Gas

Wrenn Christopher, <u>Using PIDs in Terrorist Chemical Attacks</u>, Compliance Magazine, February, 2004.

**Chemical WMD – Monitoring** Gas Monitoring Pyramid Chemical specific techniques **Colorimetric tubes Ion Mobility Spectroscopy** Gas chromatography/Mass spectroscopy PID broadband monitoring Multi-gas confined space monitors Single gas monitors Colorimetric tubes

Wrenn Christopher, <u>Using PIDs in Terrorist Chemical Attacks</u>, Compliance Magazine, February, 2004.

**Chemical WMD – Monitoring** Gas chromatography ◆ Multiple contaminants are separated into discrete peaks that are reflected on a chromatogram which facilitates identification Sample injected into column and is carried through with inert carrier gas  $(N_2, He, H_2)$ Heated column has packing or coating which attracts contaminants on various levels ◆ The greater the contaminant affinity, the slower the flow The contaminant is heated in column which, when ionized, results in formation of a peak Maslansky, Carol & Steve. Air Monitoring Instrumentation. Van Nostrand Reinhold, New York, 1993.

#### Mass spectroscopy

 Provides information about atomic and molecular composition of organic and inorganic materials; identifies unknowns and confirms presence of known or suspected contaminants

 Used in conjunction with gas chromatography, a powerful computer with an extensive analytical library containing thousands of compounds, and sample and calibration equipment in a trace air gas analysis (TAGA) mobile analytical laboratory unit.

Maslansky, Carol & Steve. <u>Air Monitoring Instrumentation</u>. Van Nostrand Reinhold, New York, 1993.

Chemical WMD – Monitoring Multi-gas confined space monitors Electrochemical sensors have coarse particulate filter, semi-permeable membrane (i.e., Teflon), an electrolyte, and electrodes ♦ Gas diffuses through membrane, dissolves in electrolyte that contains sensing electrode ♦ Gas reacts with electrode producing ions and electrons Charged particles diffuse across electrolyte to electron accepting or counting electrode Maslansky, Carol & Steve. Air Monitoring Instrumentation. Van Nostrand Reinhold, New York, 1993.

### Chemical WMD – Monitoring

### Flame Ionization Detector (FID)

- An FID uses a hydrogen-fed flame to break down organic chemicals to ions that can easily be counted with a Detector.
- Flame can ionize any organic material with IP<15.4</li>
- IP of materials NOT factor since UV lamp not used
- Instrument response depends on materials present
- FID calibrated with methane

Maslansky, Carol & Steve. <u>Air Monitoring Instrumentation</u>. Van Nostrand Reinhold, New York, 1993.

Chemical WMD – Monitoring Combustible gas monitors Wheatstone bridge – Filter-protected sensor contains two filaments: Sensing filament - treated with catalyst Compensating filament – no catalyst ◆ Battery heats both filaments to same high T<sup>o</sup> Combustible gas crossing filter and encountering both filaments is oxidized by sensing filament causing T<sup>o</sup> increase  $\diamond$  T<sup>o</sup> increase causes increase in resistance and decrease in current flow for sensing filament Maslansky, Carol & Steve. Air Monitoring Instrumentation. Van Nostrand Reinhold, New York, 1993.

### Chemical WMD – Monitoring



Selectivity Increases as you move up the Pyramid

http://www.afcintl.com/pdf/rae/ap216.pdf

### Chemical WMD – Monitoring



PID + Tubes Approximates the selectivity of GC/MS w/o the cost

http://www.afcintl.com/pdf/rae/ap216.pdf

Chemical WMD – Monitoring Commonly Used Screening Devices ♦ PIDs Less Frequently Used Screening Devices CWA detection methodologies **→** M-256 SAW Jan CAM Wrenn Christopher, Using PIDs in Terrorist Chemical Attacks, Compliance Magazine, February, 2004.

Chemical WMD – Monitoring WMD Teams Using PIDs US Army National Guard Civil Support **Teams (CST)** City HAZMAT Teams Marine Corps' CBIRF Team ◆ FBI HAZMAT Team **♦ US Secret Service** 

Christopher Wrenn is the product applications manager for RAE Systems

Wrenn Christopher, <u>Using PIDs in Terrorist Chemical Attacks</u>, Compliance Magazine, February, 2004.

### Chemical WMD – Using PPE





#### Level A

- Level A protection should be worn when the highest level of respiratory, skin, eye, and mucous membrane protection is needed.
- It consists of a fully encapsulating chemical-resistant suit and self-contained breathing apparatus (SCBA).
- This suit can be worn only for 15 to 30 minutes because the person wearing it can quickly become overheated.
- Special training is required to utilize the suit.



#### Level A

- Positive pressure (pressure demand), self contained breathing apparatus (NIOSH approved), or positive-pressure supplied air respirator with escape SCBA.
- Fully encapsulating chemical protective suit.
- Gloves, inner, chemical resistant.
- Gloves, outer, chemical resistant.
- Boots, chemical resistant, steel toe and shank; (depending on suit boot construction, worn over or under suit boot.)
- Underwear, cotton, long-john type.\*
- Hard hat (under suit).\*
- Coveralls (under suit).\*
- Two-way radio communications (intrinsically safe/non-sparking).\*

#### \* Optional

#### Level B

- Level B protection should be selected when the highest level of respiratory protection is needed but a lesser level of skin and eye protection is sufficient.
- It differs from Level A only in that it provides splash protection by use of chemicalresistant clothing (overalls, long sleeves, jacket, and SCBA).



#### Level B

- Positive-pressure (pressure-demand), self-contained breathing apparatus (NIOSH approved), or positive-pressure supplied air respirator with escape SCBA.
- Chemical resistant clothing (overalls and long-sleeved jacket, coveralls, hooded two-piece chemical splash suit, disposable chemical resistant coveralls.)
- Coveralls (under splash suit).\*
- Gloves, outer, chemical resistant.
- Gloves, inner, chemical resistant.
- Boots, outer, chemical resistant, steel toe and shank.
- Boot-covers, chemical resistant (disposable).\*
- Two-way radio communications (intrinsically safe).\*
- Hard hat. \*

■ Faceshield.\*

#### \* Optional

#### Level C

- Level C protection should be selected when the type of airborne substances is known, concentration is measured, criteria for using air-purifying respirators are met, and skin and eye exposures are unlikely.
- This involves a full-facepiece, airpurifying, canister-equipped respirator and chemical-resistant clothing.
- It provides the same level of skin protection as Level B, but a lower level of respiratory protection.



#### Level C

- Full-face or half-mask, air-purifying respirator (NIOSH approved).
- Chemical resistant clothing (one piece coverall, hooded two piece chemical splash suit, chemical resistant hood and apron, disposable chemical resistant coveralls.)
- Gloves, outer, chemical resistant.
- Gloves, inner, chemical resistant.
- Boots, steel toe and shank, chemical resistant.
- Boot-covers, chemical resistant.\*
- Cloth coveralls (inside chemical protective clothing).\*
- Two-way radio communications (intrinsically safe).\*
- Hard hat. \*

- Escape mask. \*
- Faceshield.\*

#### \* Optional



### Chemical PPE – Suit Material

#### Material

- (Type of Garment)
  - Substances Commonly Used Against
- Saranax / Tyvek SL
  - (coveralls; labcoats, sleeve protectors; aprons; hoods; Level B Suits)
    - aerosols, liquids; solvents
- **Tychem BR /; Tychem TK** 
  - (Full Level A and Level B Suits)
    - highly toxic chemicals, gases, aerosols

#### **CPF**

- (Full Level A and B suits; splash suits)
  - highly toxic chemicals; gases, aerosols

#### **PVC**

- (Full Level A suits)
  - highly toxic chemicals; gases, aerosols



### Chemical PPE – Gloves

- Glove Material
  - Resistant to:
- Viton
  - PCBs, chlorinated solvents, aromatic solvents
- Viton/Butyl
  - acetone, toluene, aromatics, aliphatic hydrocarbons, chlorinated solvents, ketones, amines and aldehydes
- **SilverShield and 4H (PE/EVAL)** 
  - morpholine, vinyl chloride, acetone, ethyl ether, many toxic solvents and caustics
- Barrier
  - ◆ Wide range of chlorinated solvents, aromatics, acids
- **PVA** 
  - ketones, aromatics, chlorinated solvents, xylene, MIBK, trichloroethylene; DO NOT USE WITH WATER /AQUEOUS SOLUTIONS
- Butyl
  - aldehydes, ketones, esters, alcohols, most inorganic acids, caustics, dioxane



### Chemical PPE – Gloves

#### Glove Material

• Resistant to:

#### Neoprene

- oils, grease, petroleum -based solvents, detergents, acids, caustics, alcohols, solvents
- **PVC** 
  - ◆ acids, caustics, solvents, solvents, grease, oil
- Nitrile
  - ♦ oils, fats, acids, caustics, alcohols
- Latex
  - ◆ body fluids, blood, acids, alcohols and alkalis
- Vinyl
  - ♦ body fluids, blood, acids, alcohols and alkalis
- **Rubber** 
  - organic acids, some mineral acids, caustics, alcohols; not recommended for aromatic solvents, chlorinated solvents



Chemical PPE – Respirators
Atmosphere-Supplying Respirators:

Self-Contained Breathing Apparatus (SCBA)
 Closed circuit

 Compressed or Liquid Oxygen; Oxygen Generating
 Open Circuit
 Demand; Pressure-Demand; Continuous Flow

 Supplied Air Respirators (SAR)

 Hose Mask
 With or Without Blower

- Airline Respirator
  - Demand; Pressure-Demand; Continuous Flow
- Combination Airline Respirator with auxiliary air supply



# Chemical PPE – SCBA Respirator Self-Contained Breathing Apparatus (SCBA):

#### General Advantages

- Provides highest available level of protection against airborne contaminants and oxygen deficiency
- Provides highest available level of protection under strenuous work conditions
- General Disadvantages
  - Bulky, heavy (up to 35 pounds)
    Finite air supply limits work duration
    May impair movement in confined spaces



Chemical PPE – SCBA Respirator
Open Circuit (SCBA): ENTRY & ESCAPE

#### Advantages

Warning alarm sounds when 20-25% of air supply remains

- Disadvantages
  - Bulky, heavy (up to 35 pounds)

Finite air supply (30-60 min.) limits work duration

Size of tank and work rate of individual vary operating time



## Chemical PPE – SCBA Respirator

- Closed Circuit (SCBA): ENTRY & ESCAPE
- Advantages
  - Warning alarm sounds when 20-25% of air supply remains
  - Longer operating time (up to 4 hrs.)
  - $\diamond$  O<sub>2</sub> depleted before CO<sub>2</sub> supply; no CO<sub>2</sub> breakthrough
- Disadvantages
  - $\diamond$  CO<sub>2</sub> break through may occur at very cold T<sup>o</sup>
  - ♦ Unit retains heat from exhalation heat stress
  - When worn outside of suit contamination of breathing bag possible



Chemical PPE – SCBA Respirator **ESCAPE ONLY (SCBA):** 

Advantages ◆Light weight (10 lbs.), low bulk Available in pressure demand and continuous flow modes Disadvantages Cannot be used for entry Provides only 5-15 min of protection depending on model/breathing rates



### Chemical PPE – SAR Respirator

Supplied Air Respirators (SAR):

#### Advantages

Enables longer work periods than an SCBA

- ◆ Less bulky and heavy than SCBA (5-15 lbs.)
- Protects against most airborne contaminants
- Disadvantages
  - $\diamond$  Not approved for IDLH or O<sub>2</sub> deficient areas
  - Impairs mobility
  - ♦ MSHA/HIOSH hose length limited to 300 feet
  - ♦ Length of hose may be compromised by:
    - Air flow, damage, contamination, degradation
    - Processity to retrace steps to exit environment
    - Necessity of air supply line supervision and monitoring


Chemical PPE – APR Respirator Atmosphere-Purifying Respirators (APRs):

Advantages Enhanced mobility ◆Lighter in weight than a SCBA (<2 lbs.) Disadvantages  $\diamond$  Cannot be used in IDLH or O<sub>2</sub> deficient area ◆ Limited duration of protection (indicator?) Chemical and concentration specific  $\diamond$  Requires monitoring of contaminant/O<sub>2</sub> levels Contaminants must have adequate warning properties or trigger end of service indicator

## Radiological Types of WMD



#### Radiological Types of WMD

"Dirty" Bombs – Consists of a conventional explosive that is packaged with radioactive material that scatters when bomb explodes

Risk would only be to immediate area of explosion and can be remediated in a short time.

#### Radiological Types of WMD

- Note: No instrument can measure all types of radiation at all levels of exposure
- Quickly move away from immediate area, at least several blocks (reduce inhalation of radiation)
- Tune in to local radio or TV broadcasts for more info
- Potassium iodide not necessarily protective because radioactive iodine not necessarily isotope used in device and is only specific to thyroid; must be taken before or right after exposure

Radiological Types of WMD The three basic ways to reduce radiation exposure are through:

■ TIME

Decrease the amount of time you spend near the source of radiation.

DISTANCE

Increase your distance from a radiation source.
 SHIELDING

Increase the shielding between you and the radiation source.

Radiological Terrorism Tips: AIHA Synergist - December 2003 (University/Hospital Site)

Personnel who can neither hear nor see an explosion are probably not at risk.

- Stay put if indoors
- If outdoors, go inside to await further information and instructions

Driving away is discouraged, because driving is likely to be more dangerous than staying put

# Radiological Terrorism Tips

After going indoors, personnel should:

 close open doors and windows
 wash their hands and face (taking a shower if possible)
 change their outer clothes if they can

- Contaminated injured people should have serious injuries treated without regard to contamination levels—they do not endanger emergency response or medical personnel
  - If injuries are not serious, it may be possible to decontaminate the victims before transporting, or wrap them in a sheet to minimize contamination
  - Depends on the extent of injuries and contamination

## Radiological Terrorism Tips

- You may need to perform surveys to establish radiological boundaries for high radiation or high contamination levels.
  - Radiation surveys are relatively easy to perform, and boundaries often established fairly easily
  - Contamination boundaries are more difficult to establish because surveys can be difficult and timeconsuming

According to regulations:

- the limit for removable contamination in an unrestricted area is 1,000 dpm/100 cm<sup>2</sup>
- radiation levels in uncontrolled areas cannot exceed 2 mrem in one hour.

#### Radiological Terrorism Tips

 In some cases, it may be best to simply set contamination control boundaries a few hundred meters downwind and then expand or collapse them as you survey to confirm them.

#### **Radiological Survey Tips**

■ Be "low and slow": The detector should be:

- held no more than a half inch (1 cm) from the surface being surveyed (avoid contamination and false "low")
   moved no more quickly than 1–2 inches per second (3–
  - 5 cm/second). (avoid missing contaminated area)
- While surveying, keep the audible response turned on so you can watch the probe itself.
- Listen to the count rate:
  - If you hear an increase, pause for a moment to see if the increase is sustained
  - If so, look at the meter face to see what the count rate is at that location.

#### Radiological Survey Tips

- When logging results, and in decontamination, you must convert from counts per minute (what the meter reads out) to disintegration per minute (the amount of contamination present).
- Divide the count rate by the meter efficiency, which should be determined when the meter is calibrated.
- For example, if a Geiger counter has 40 percent percent detection efficiency for P-32, a count rate of 80 cpm above background levels corresponds to a disintegration rate of 200 dpm (80 ÷ 0.4 = 200).

## Radiological Survey Tips

#### Table 1: How to Choose a Survey Meter

Type of Radiation Emitted	Example Isotopes	Type of Survey	Type of Detector to Use
Alpha	U-238, Pu-238, Pu-239,	Direct frisk,	Zinc sulfide (ZnS) or
	Ra-226, Po-210, Am-241	smear wipe	proportional counter
Low-energy beta	H-3, C-14, S-35, Pu-241	Smear wipe	Liquid scintillation counter, proportional counter
Medium- to	P-32, Sr-90, I-131	Direct frisk,	Geiger counter, liquid scintillation
high-energy beta		smear wipe	counter, proportional counter
Low-energy gamma	I-125, I-129, Am-241	Direct frisk or smear wipe	Thin-crystal (1″ × 1 mm) sodium iodide
Medium- to	I-131, Cs-137, Co-60, Ir-192	Direct frisk or	Thick-crystal (1″ × 1″ or larger)
high-energy gamma		smear wipe	sodium iodide, Geiger counter

Andrew Karam is with Rochester Institute of Technology, Rochester, N.Y. (AIHA Synergist, Volume 14, Number 12, December 2003)